

An optical characterization of Tris(2-aminoethyl)aminocarboxylic acid**Functionalized Polyvinylbenzyl Chloride Microspheres using polymer swelling**

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Abstract

In this work, Tris(2-aminoethyl)aminocarboxylate functional group was attached chemically to the backbone of polyvinylbenzyl chloride which was lightly cross-linked with divinyl benzene. This ligand has a chemical structure similar to that of ethylenediamine tetraacetic acid (EDTA), as a common chelating agent for metal ions. These derivatized polymer microspheres were embedded in a hydrogel matrix of poly vinyl alcohol cross-linked with glutaraldehyde, which is optically characterized. The response is based on the interaction between the metal cations with the negative charges of the deprotonated carboxylate functional group, which led to neutralization of the charges and thus to polymer shrinking. As a result, an increase in the turbidity of the sensing membrane due to a change in the refractive index between the hydrogel and the derivatized polymer microspheres occurred. The changes in the turbidity of the sensing membrane were measured as absorbance using a conventional spectrophotometer. This sensing membrane responded selectively to Cu²⁺ solutions of different concentrations range (1x10⁶M to 1x10⁴ M). It is found that Cu²⁺ bind to the demonstrated functionalized-polymer with formation constant of 1 x 10⁵ M⁻¹.

References:

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