

# Poster Presentations

## Pure Palladium nanoparticles through themolysis of $\text{PdCl}_2(\text{Pan})$ complex under microwave Mode

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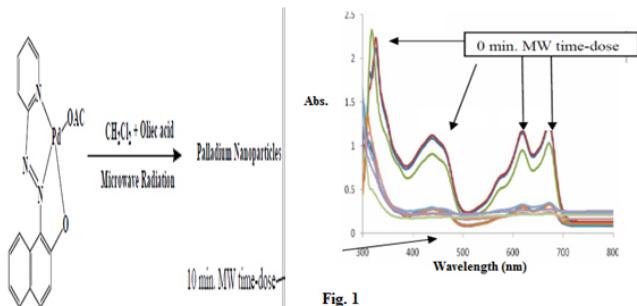
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### Abstract

$\text{PdCl}_2(\text{Pan})$  complex was served as precursor to prepare uniform and stable Palladium nanoparticles using oleic acid as solvent and reducing agent under time-depended microwave radiation (from 0-10 min). Reduction of  $\text{PdX}_2(\text{Pan})$  by MW to Pd nanoparticles was monitored by UV-visible spectroscopy at four different maxima wavelengths as depicted in Figure 1.



The new Pd-nanoparticle material were characterized by several physical methods, such as UV-vis spectroscopy, IR, NMR, SEM, TEM, XRD analysis and PSA which confirmed the formation of nano-palladium material. Both the pure Palladium complex and Palladium NPS revealed high catalytic activity when subjected to Heck C-C cross coupling reaction under basic mild conditions [1-4].

### References

- [1] C. Burda, X. Chen, R. Narayanan, M.A. El-Sayed, Chem. Rev. 105 (4) (2005) 1025– 1102.
- [2] W.W. Yu, H. Liu, J. Mol. Catal. A: Chem. 243 (2006) 120–141.
- [3] P. Moriarty, Rep. Prog. Phys. 64 (2001) 297–381.
- [4] R.F. Ziolo, E.P. Giannelis, B.A. Weinstein, M.P. O'Horo, B.N. Ganguly, V. Mehrotra, M.W. Russell, D.R. Huffman, Science 257 (1992) 219–223.