## **Oral Presentation**

## Biocompatible Nanoemulsion : phase behavior ,Formulation, Characterization and some application

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## Abstract

Isotretinoin is 13-cis-retinoic acid and is related to both retinoic acid and Retinol (Vitamin A). It has been commonly used for the treatment of severe acne and the other dermatological diseases, isotretinoin has some deficiencies, such as poor solubility in water and in most organic solvents and poor stability, being easily oxidized when heated or exposed to light.

Because water insoluble drugs often show low absorption and weak bioavailability, improvement in solubility is important for development of drug preparations.

Drugs can be solubilized and formulated in nanoemulsions . nanoemulsions are excellent candidates as potential drug delivery systems because of their improved drug solublization , long shelf life and ease of preparation .

Recently, Tetronic surfactants have been studied as possible vehicles for drug delivery; hence, studies on their behavior under a variety of conditions will be an important part of the formulation in delivery agents.

Tetronic 1107 is a tetrafunctional block copolymer surfactant terminating in primary hydroxyl groups see figure (1). A nonionic surfactant that is 100% active and nontoxic

This study aims to investigate the phase behavior of Tetronics 1107 with Propylene Glycol as a model oil and cationic surfactant tetra butyl ammonium bromide at different temperature (25,37,and 45 C), and then investigate the phase behavior of Tetronics 1107 with R (+)-Limonene oil at different temperature (25,37C) to form nanoemulsion in order to improve solubility of isotretinoin

Visual inspection, cross polarizers and polarized microscopy were used to detect anisotropy. A cubic phase and micelle were detected in the corresponding ternary phase diagram, each of them will be used to formulate of isotretinoin in a second stage.

$$\begin{array}{c} \text{HO(EO)}_{60}(\text{PO)}_{20} \\ \text{HO(EO)}_{60}(\text{PO)}_{20} \end{array} \text{N-CH}_2\text{-CH}_2\text{-N} \\ \begin{array}{c} \text{(PO)}_{20}(\text{EO)}_{60}\text{OH} \\ \text{(PO)}_{20}(\text{EO)}_{60}\text{OH} \end{array}$$

Figure (1): chemical structure of tetronics 1107